

# **SRF LIMITED**

Issue: 02 Rev: 01 Revision Date: 01.07.2020

# SAFETY DATA SHEET

**R410A** 

#### SECTION 1: IDENTIFICATION OF SUBSTANCE OR MIXTURE AND COMPANY

1.1 Product Name : FLORON R-410A

Trade Names / Synonyms : F410A, R410A, BLEND GAS

Ingrediant	CAS number	Weight
Difluoromethane (R32)	75-10-5	50
Pentafluoroethane (R125	354-33-6	50

**1.2** Manufacturer/supplier : SRF Limited, D-2/1 GIDC Phase-II, PCPIR, Dahej, Tal. Vagra,

Dist. Bharuch 392 130, Gujarat (India)

Further information obtainable from: : Prabhat Kumar

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1.3 Emergency Call

**CAS Number** 

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 : Mr. Sharma Anil Kumar
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Relevant Identified Uses Of The Substance Or Mixture And Uses Advised Against

Identified Uses: : Refrigerant

**Uses advised against:** : Do not use product for anything outside of the above specified

uses

## **SECTION 2: HAZARDS IDENTIFICATION**

## 2.1 Classification of the substance or mixture :

#### **Emergency Overview**

Colorless, volatile liquid with ethereal and faint sweetish odor. Non-flammable material. Overexposure may cause dizziness and loss of concentration. At higher levels, CNS depression and cardiac arrhythmia may result from exposure. Vapors displace air and can cause asphyxiation in confined spaces. At higher temperatures, (>250degC), decomposition products may include Hydrofluoric Acid(HF) and carbonyl halides

Physical Hazards : H280 Contains gas under pressure; may explode if heated

Other hazards : May displace oxygen and cause rapid suffocation

: Vapours are heavier than air and can cause suffocation by reducing oxygen available for breathing. Misuse or intentional inhalation abuse may cause death without warning symptoms,

due to cardiac effects.

Rapid evaporation of the product may cause frostbite

Precautionary statement(s)

Prevention : P202 Do not handle until all safety precautions have been read

and understood.

P244 Keep valves and fittings free from oil and grease.P217 Use only outdoors or in a well-ventilated area



Response : P308+P313 IF exposed or concerned: Get medical

advice/attention.

: P303+P361+P353 IF ON SKIN (or hair): Take off Immediately all contaminated clothing. Rinse SKIN with water or shower.

: P304+P341 IF INHALED: If breathing is difficult, remove victim to fresh air and keep at rest in a position comfortable for

breathing.

: P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses if present and easy to

do - continue rinsing

: P371+P380+P375 In case of major fire and large quantities: Evacuate area. Fight fire remotely due to the risk of explosion.

: P410+P403 Protect from sunlight. Store in a well-ventilated

place.

**Disposal** : P 501 Dispose of contents/container in accordance with all local,

regional, national and international regulations

Label elements

Storage

Labelling according Regulation (EC) No 1272/2008:

Pictogram :

 $\Diamond$ 

Signal Word : Warning

**Supplemental label information**: EIGA-0783: Contains fluorinated greenhouse gases covered by

the Kyoto protocol.

EIGA-As: Asphyxiant in high concentrations

# **SECTION 3: COMPOSITION & INFORMATION ON INGREDIENTS**

Chemical name	Common name and synonyms	Formula	CAS No.	Concentration % (w/w)
Difluoromethane	Methylene fluoride; Carbon fluoride hydride (CF2H2); Difluoromethane; Freon 32, R32	CH2F2	75-10-5	50
Pentafluoroethane	Refrigerant gas F125; R125, Halocarbon 125	CHF2CF3	354-33-6	50

#### **SECTION 4. FIRST AID MEASURES**

# 4.1 Description of first aid measures General advice

: In high concentrations may cause asphyxiation. Symptoms may include loss of mobility/consciousness. Victim may not be aware of asphyxiation. Remove victim to uncontaminated area wearing self contained breathing apparatus. Keep victim warm and rested. Call a doctor. Apply artificial respiration if breathing stopped.

In the case of accident or if you feel unwell, seek medical advice immediately. When symptoms persist or in all cases of doubt seek medical advice



**Inhalation** : In high concentrations may cause asphyxiation. Symptoms

may include loss of mobility/consciousness. Victim may not be aware of asphyxiation. Remove victim to uncontaminated area wearing self contained breathing apparatus. Keep victim warm and rested. Call a doctor. Apply artificial respiration if breathing

stopped. Get medical attention if symptoms occur.

**Skin contact**: Contact with evaporating liquid may cause frostbite or freezing

of skin. Treat for frostbite if necessary by gently warming affected area. Do not rub affected area. Get medical attention

immediately.

**Eye contact:** : In case of contact, immediately flush eyes with plenty of water

for at least 15 minutes. Get medical attention immediately

**Ingestion:** : Ingestion is not considered a potential route of exposure.

4.2 Most Important Symptoms And Effects, Both Acute And Delayed:

: Anaesthetic effects Light-headedness irregular heartbeat with

a strange sensation in the chest, heart thumping, apprehension, feeling of fainting, dizziness or weakness.

4.3 Indication of any immediate medical attention and special treatment needed

**Hazards:** : Respiratory arrest. Contact with liquefied gas can cause

damage (frostbite) due to rapid evaporative cooling.

**Treatment:** : Thaw frosted parts with lukewarm water. Do not rub affected

area. Get immediate medical advice/attention

**Protection of first-aiders**: : If potential for exposure exists refer to Section 8 for specific

personal protective equipment.

**Notes to physician**: Treat symptomatically and supportively.

Because of possible disturbances of cardiac rhythm, catecholamine drugs, such as epinephrine, that may be used in situations of emergency life support should be used with special caution. Avoid administration of adrenaline or other

simpatomimeticas similar, as it can produce a cardiac

arrhythmia with possible later heart failure.

#### **SECTION 5. FIRE FIGHTING MEASURE**

Fire and explosion hazards

: R410A (HFC410A) is not flammable in air under ambient conditions of temperature and pressure. Under conditions of high temperature and pressure, certain HFC125 (F125) /air mixtures were shown to be flammable. Certain mixtures of HFC125 (F125) and chlorine may be flammable under some conditions.



- 5.1 Extinguishing media Suitable extinguishing media:
- 5.2 Special hazards arising from the substance or mixture

**Hazardous Combustion Products:** 

Advice for firefighters Special fire fighting procedures:

Special protective equipment for firefighters:

- : Containers may burst under intense heat. Ruptured cylinders may rocket or fragment. Heavy vapour may suffocate
- : Use extinguishing measures that are appropriate to local circumstances and the surrounding environment. (Material itself is not flammable)
- : Cylinders are equipped with pressure and temperature relief devices, but may still rupture under fire conditions.

  Decomposition may occur. Contact of welding or soldering torch flame with high concentrations of this substance can result in visible changes in the size and colour of the torch flame. This flame effect will only occur in concentrations of this substance well above the recommended exposure limit.

  Therefore stop all work and ventilate to disperse vapours from the work area before using any open flames.

  This substance is not flammable in air at temperatures up to

100 deg. C (212 deg. F) at atmospheric pressure. However, mixtures of this substance with high concentrations of air at elevated pressure and/or temperature can become combustible in the presence of an ignition source. This substance can also become combustible in an oxygen enriched environment (oxygen concentrations greater than that in air). Whether a mixture containing this substance and air, or this substance in an oxygen enriched atmosphere become combustible depends on the inter-relationship of 1) the temperature2) the pressure, and 3) the proportion of oxygen in the mixture. In general, this substance should not be allowed to exist with air above atmospheric pressure or at high temperatures; or in an oxygen enriched environment. For example, this substance should NOT be mixed with air under pressure for leak testing or other purposes. Experimental data have also been reported which indicate combustibility of this substance in the presence of certain concentrations of chlorine

- : If involved in a fire the following toxic and/or corrosive fumes may be produced by thermal decomposition: carbon monoxide ; Carbonyl difluoride ; Hydrogen fluoride
- : In the event of fire, wear self-contained breathing apparatus.

  Use personal protective equipment. Wear neoprene gloves
  during cleaning up work after a fire. Exposure to decomposition
  products may be a hazard to health.
- : In case of fire: Stop leak if safe to do so. Continue water spray from protected position until container stays cool. Use extinguisher to contain the fire. Isolate the source of the fire or let it burn out. In case of major fire and large quantities: Evacuate area. Fight fire remotely due to the risk of explosion.

Evacuate area. Fight fire remotely due to the risk of explosion. Fire-fighters must use standard protective equipment including

flame retardant coat, helmet with face shield, Gloves, rubber boots, and in enclosed spaces, SCBA.

# SECTION 6: ACCIDENTAL RELEASE MEASURES



6.1	Personal precautions, protective	:	Evacuate area. Provide adequate ventilation. Prevent from
	equipment and emergency procedures:		entering sewers, basements and workpits, or any place where
			its accumulation can be dangerous. Wear self-contained
			breathing apparatus when entering area unless atmosphere is

proved to be safe.

**6.2** Environmental Precautions : Prevent further leakage or spillage if safe to do so.

6.3 Methods and material for containment : Provide adequate ventilation.and cleaning up : Avoid open flames and high temperatures

**6.4 Reference to other sections:** : Refer to sections 8 and 13.

#### **SECTION 7: HANDLING & STORAGE**

# 7.1 Precautions for safe handling:

: Only experienced and properly instructed persons should handle gases under pressure.

Handle in accordance with good industrial hygiene and safety practice.

Protect containers from physical damage; do not drag, roll, slide or drop. Do not remove or deface labels provided by the supplier for the identification of the container contents. When moving containers, even for short distances, use appropriate equipment eg. trolley, hand truck etc.

Secure cylinders in an upright position at all times, close all valves when not in use. Provide adequate ventilation. Use a check valve or trap in the discharge line to prevent hazardous back flow into the cylinder.

When using do not eat, drink or smoke.

Use a pressure reducing regulator when connecting cylinder to lower pressure (<3000 psig) piping or systems. Never attempt to lift cylinder by its cap. Never use direct flame or electrical heating devices to raise the pressure of a container. Leave valve protection caps in place until the container has been secured against either a wall or bench or placed in a container stand and is ready for use. Damaged valves should be reported immediately to the supplier Close container valve after each use and when empty, even if still connected to equipment. Never attempt to repair or modify container valves or safety relief devices. Replace valve outlet caps or plugs and container caps where supplied as soon as container is disconnected from equipment. Keep container valve outlets clean and free from contaminates particularly oil and water. If user experiences any difficulty operating container valve discontinue use and contact supplier. Never attempt to transfer gases from one container to another. Container valve guards or caps should be in place.

The product should not be mixed with air for leak testing or used with air for any other purpose above atmospheric pressure. Contact with chlorine or other strong oxidizing agents should also be avoided.

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7.2 Conditions for safe storage, including any incompatibilities:

Observe all regulations and local requirements regarding storage of containers.

Protect from sunlight. Store in a well-ventilated place. Containers should not be stored in conditions likely to encourage corresion. Stored containers should be per

encourage corrosion. Stored containers should be periodically checked for general conditions and leakage. Container valve guards or caps should be in place. Store containers in location free from fire risk and away from sources of heat and ignition.

Keep away from combustible material.

**7.3 Storage temperature** : Protect from sunlight. Store in a cool and well-ventilated place.

**7.4 Storage period:** : No data available

#### **SECTION 8: EXPOSURE CONTROL / PERSONAL PROTECTION CONTROL PARAMETERS**

# 8.1 **Control parameters**

Expossure guidline

Ingrediant Name	ACGIH TLV	OSHA PEL	Other Limit	
Difluoromethane	None	None	# 1000 ppm TWA	
Pentafluoroethane	None	None	# 1000 PPM TWA	
#= Workplace envoirmental Expossure level (AIHA)				

#### 8.2 Exposure controls

Engineering controls

: Normal ventilation for standard manufacturing procedures is generally adequate. Local exhaust should be used when large amounts are released. Mechanical ventilation should be used in low or enclosed places. Concentration monitors may be necessary to determine vapour concentrations in work areas prior to use of torches or other open flames, or if employees are entering enclosed areas.

Personal protective equipment

Respiratory protection

: For rescue and maintenance work in storage tanks use selfcontained breathing apparatus. Vapours are heavier than air and can cause suffocation by reducing oxygen available for

breathing.

**Hand protection** 

: Additional protection: Wear approved gloves that are suitable for the task and have been shown to be impervious for the

duration of their use

Eye protection

: Wear safety glasses with side shields. Additionally wear a face shield where the possibility exists for face contact due to

**Protective measures** 

splashing, spraying or airborne contact with this material.When using do not smoke. Self-contained breathing apparatus

(SCBA) is required if a large release occurs

Other exposure limits for potential

decomposition prosucts

: Hydrogen Fluoride: ACGIH TLV: 2 ppm ceiling,



#### **SECTION 9: PHYSICAL & CHEMICAL PROPERTIES**

# 9.1 Information on basic physical and chemical properties

Appearance : Clear, colourless liquified gas

Physical state : Liquified gas
Colour : colourless

Odor : Faint ethereal odor Odor Threshold : No data available

Molecular Weight : 72.6

Chemical Formula :CH2F2, CHF2CF3

pH (15 aqueous solution) : Neutral

Melting point/freezing point : Not Determined

Initial boiling point and boiling range : -48.5 °C (-55.3 °F) at 1,013 hPa

Flash point : None (Does not Flash)
Flammability : No data available
Upper/lower flammability or explosive limits : Does not flash

Ozone Depletion Potential : Nil (Source: as per schedule- 1 of ozone legislation ,Ozone

cell-Management of environment forest & climate change)

Global Warming Potential : 2088 (Source:IPCC-AR4 Report)

Vapour pressure : 215.3 psia at 70°F
Vapour density : 3.0 at 25°C (Air = 1.0)
Water solubility : No Data available
Auto-ignition temperature : No data available
Decomposition temperature : No data available

#### **SECTION 10: STABILITY & REACTIVITY**

**10.1 Reactivity:** : Decomposes on heating.

10.2 **Chemical stability** : Stable under recommended storage conditions

10.3 **Possibility of hazardous reactions** : Polymerization will not occur

10.4 **Conditions to avoid:** : The product is not flammable in air under ambient

conditions of temperature and pressure. Do not mix with oxygen or air above atmospheric pressure. Any source of high temperature, such as lighted cigarettes, flames, hot

spots or welding may yield toxic and/or corrosive

decomposition products.

10.5 **Incompatible materials:** : Alkali metals Alkaline earth metals, Powdered metals,

Powdered metal salts

10.6 **Hazardous decomposition products**: Decomposition products are hazardous., This material can be

decomposed by high temperatures (open flames, glowing metal surfaces, etc.) forming hydrofluoric acid and possibly carbonyl fluoride., These materials are toxic and irritating.,

Avoid contact with decomposition products.

# **SECTION 11: TOXICOLOGICAL INFORMATION**



# 11.1 Information on toxicological effects\

Difluoromethane	LC50 : 4 hr. (rat) - > 520,000
Pentafluoroethane	Cardiac Sensitization threshold (dog) > 100,000 ppm

Carcinogenicity : Not classifiable as a human carcinogen. Overall weight of

evidence indicates that the substance is not carcinogenic

**Mutagenicity** : Animal testing did not show any mutagenic effects. Tests on

bacterial or mammalian cell cultures did not show mutagenic

effects

**Reproductive toxicity**: No toxicity to reproduction

No effects on or via lactation

Animal testing showed no reproductive toxicity

**Teratogenicity**: Animal testing showed no developmental toxicity

# 12. ECOLOGICAL INFORMATION

Degradability (BOD): R-410A is a gas at room temperature; therefore, it is unlikely to remain in water.

#### **SECTION 13: DISPOSAL CONSIDERATIONS**

#### 13.1 Waste Treatment Methods

Product : Dispose in accordance with all applicable regulations. Comply

with applicable Federal, State/Provincial and Local Regulations

Contaminated packaging : Empty pressure vessels should be returned to the supplier.

#### SECTION 14: TRANSPORT INFORMATION

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ITEM	ADR	IMDG	IATA
UN number	3163	3163	3163
Proper shipping name	LIQUEFIED GAS, N.O.S. (Pentafluoroethane, Difluoromethane)	LIQUEFIED GAS, N.O.S. (Pentafluoroethane, Difluoromethane)	LIQUEFIED GAS, N.O.S. (Pentafluoroethane, Difluoromethane)
Transport hazard class(es)/ Labelling Number	2.2	2.2	2.2
Packaging Instruction	P 200`	P 200	P 200
Environmental hazards	No	No	No
Special precautions for user	No data is available on this product		



# 15. REGULATORY INFORMATION

#### U.S. Federal Regulations

None of this product's components are listed under SARA Section 302/304 (40 CFR 355 Appendix A), SARA Section 311/312 (40 CFR 370.21), SARA Section 313 (40 CFR 372.65), CERCLA (40 CFR302.4), TSCA 12(b), or require an OSHA process safety plan.

SARA 311/312 Acute: Yes Chronic: No Fire: No Reactive: No Pressure: Yes

### U.S. State Regulations

None of this product's components are listed on the state lists from CA, MA, MN, NJ, PA or

RI.

Not regulated under California Proposition 65

# **SECTION 16: OTHER INFORMATION**

The above information is believed to be correct but does not purport to be all inclusive and shall be used only as a guide. The information in this document is based on the present state of our knowledge and is applicable to the product with regard to appropriate safety precautions. It does not represent any guarantee of the properties of the product. SRF Limited-Chemical business shall not be held liable for any damage resulting from handling or from contact with the above product

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